

Global Circulation of Carbon related to Climate Change and Environment

<http://timetraveler.html.xdomain.jp>

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Climate change, population increase, and food problem

- World population will increase to 20 billion in 2050.
- The increase in food production to match the increased population can not be expected due to the global warming and climate change.
- Big typhoon → Flooding
- El Nino → Drought
- Salt accumulation in the crop land
 - ← Flooding in the coastal area
 - ← Salt accumulation due to drought

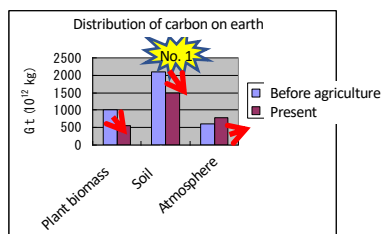
Large amount of gas is emitted from soil surface



Global Warming Potential

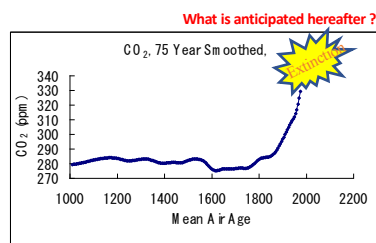
| | Gasses | GWP |
|---|---|--------|
| 1 | Carbon dioxide (CO ₂) | 1 |
| 2 | Methane (CH ₄) | 21 |
| 3 | Nitrous oxide (N ₂ O) | 310 |
| 4 | Trifluoromethane (CHF ₃) | 11,700 |
| 5 | Difluoromethane (CH ₂ F ₂) | 650 |
| 6 | Fluoromethane (CH ₃ F) | 150 |

Distribution of C on Earth



Organic matter in plant and soil decreased remarkably due to human civilization.

土壤生化学 (1994)



Change in atmospheric CO₂ concentration. (From the Antarctic ice core data.)

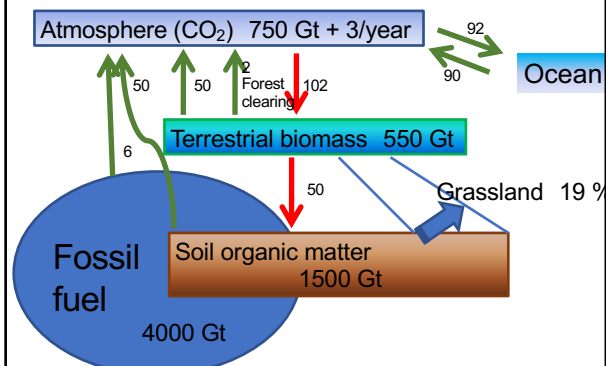
土壤学概論(2001) 表7.1 で紹介

Stocks of carbon on the surface of earth

| Stock pools | Stored amount (Gt=10 ¹² kg) |
|--------------------------------|---|
| Earth | |
| Plant biomass | 550 |
| Soil humus | 1500 |
| Atmosphere | 1850 (CO ₂ 260 ppm) |
| | 1890 (CO ₂ 290 ppm) |
| | 2000 (CO ₂ 390 ppm) |
| Ocean | 38000 |
| Carbonate salts | 20x10 ⁶ |
| Dissolved organic matter | 600 |
| Solid suspension and sediments | 3000 |
| Earth crust (fossil fuel) | 4000 |
| Total amount | 44800 |

Hunt(1972), Paul and Clark(1989), Eswaran et al.(1993)
CO₂ concentration was calculated from ice-core data in Law Dome Antarctica.

Terrestrial carbon pool and its flux.



Occurrence of Nitrogen on Earth and its pool size.

| Occurrence | 10 ⁶ t |
|-------------------------------|------------------------|
| Atmosphere | 3.9 × 10 ⁹ |
| Terrestrial Plant | 15 × 10 ³ |
| Animal | 0.2 × 10 ³ |
| Soil organic matter | 150 × 10 ³ |
| Ocean Biomass | 0.5 × 10 ³ |
| Soluble and sediment | 1200 × 10 ³ |
| Nitrate nitrogen in the above | 570 × 10 ³ |

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Occurrence of Phosphorus on Earth and its pool size.

| Occurrence | 10 ⁶ t |
|---------------------|-----------------------------|
| Terrestrial Biomass | 2.6 × 10 ³ |
| Phosphorus rock | 19 × 10 ³ |
| Soil | 96~160 × 10 ³ |
| Fresh water | 0.090 × 10 ³ |
| Ocean Biomass | 0.05~0.12 × 10 ³ |
| Soluble inorganic P | 80 × 10 ³ |
| Sediment | 840,000 × 10 ³ |

Soil is the largest stock for C, N, P in the terra.

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Biomass production and Respiration/combustion on earth (10⁹ t C/year)

| | Biomass | CO ₂ Production |
|-----------|---------|----------------------------|
| Plant | 500 | 34.5 |
| Animal | 0.5 | 4.1 |
| Human | 0.1 | 0.7 |
| Microbes | 1.0 | 112 |
| Fire | | 6.9 |
| Eruption | | 0.15 |
| Factories | | 15 |
| Total | 502 | 173.5 |

Energy consumption by 1 person

- World (Average) 1.7 t /year (petrol equivalent)
- Japan 4.1 t /year
- USA 8.0 t /year
- Human life increases the atmospheric CO₂ concentration.
- Plant and Soil absorb and store the emitted carbon.

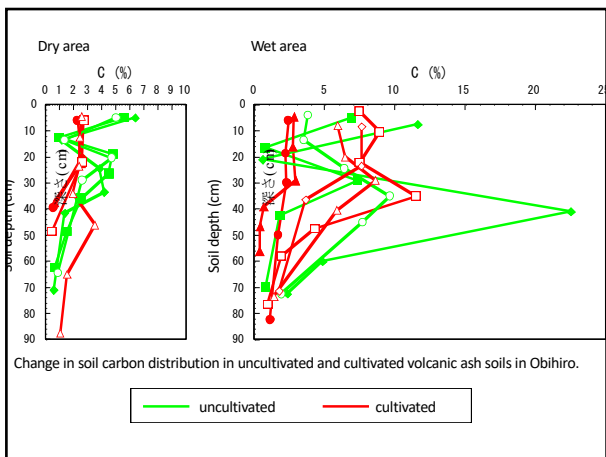
World energy consumption (2003)

| Source | Consumption (petroleum equivalent 10 ⁸ tons) | |
|-------------|---|------|
| Petroleum | 36.4 | 85.5 |
| Natural gas | 23.3 | |
| Coal | 25.8 | |
| Atomic | 6.0 | 12.0 |
| Hydraulic | 6.0 | |

CO₂ emission (includes Petroleum, Natural gas, Coal)
heat emission (includes Atomic, Hydraulic)

Emission of CO₂ by human.

| Factors | Increasing rate of CO ₂ |
|----------------------------------|------------------------------------|
| | Gt (10 ⁹ t)/year |
| Combustion of fossil fuel | 7 |
| Land use change | 2.2 |



Greenhouse gas emission in Japan (2017).

| Gases | (10 ⁶ t) | Items | (10 ⁶ t CO ₂ equivalent.) | | |
|-----------------|---------------------|------------------|---|--|---|
| CO ₂ | total | 1191 | | | |
| | Energy origin | 1112 | industry 413, transportation 213, service 206, domestic energy transformation 188, energy transformation 92.3 | | |
| | Non-energy | 79.3 | industrial process 46.2, wastes combustion 29.8, agriculture 3.3 | | |
| | CH ₄ | total | 30.5 | agriculture 23.5, waste treatment 4.9, fuel combustion 1.3 | |
| | | N ₂ O | total | 20.5 | agriculture 9.5, fuel combustion, leak 6, waste treatment 4 |
| | | | Hydrofluorocarbon | 45.7 | 45.7 |
| Perfluorocarbon | 3.4 | 3.4 | | | |

Data from the Ministry of Environment, Japan (2018). <https://www.env.go.jp/press/106211.html>

Emissions due to agriculture

| Gasses | land use change | forest clearing | Countermeasures |
|--------------------|------------------|-----------------------------|--|
| CO ₂ | land use change | forest clearing | Stop or decrease forest clearing. |
| | | grassland turning | Stop turning grassland to cropland. |
| | | peatland burn and drain | Stop agricultural use of peatland. Do not drain the peatland. |
| machine operation | fuel consumption | | Decrease the frequency of machine use. |
| agricultural waste | burning | | Do not burn the crop residue. |
| | | | Recycle the agricultural waste. |
| soil | soil respiration | | Minimise ploughing or non-ploughing. |
| | ploughing | | Return organic matter and animal excreta to soil after composting. |
| | | | Grow green manure. |
| CH ₄ | agriculture | paddy field | Do not apply fresh organic matter. |
| | | domestic animals | Intermittent drying of paddy field. |
| N ₂ O | agriculture | N fertilizer transformation | Decrease the use of inorganic fertilizer. |
| | | denitrification | Do not make the anaerobic soil condition. Grow legume green manure for N source. |

Stabilization and abundance of organic matter constituents in soil

| Constituents | Abbreviation | Mean Residence Time | S (kg) | A ⁰ (kg) |
|--------------------------------------|--------------|---------------------|------------------------|---------------------|
| Fresh organic matter (yearly input) | | | | 1000 |
| Decomposable Plant Material | DPM | 1 | 10 | 10 |
| Refractory Plant Material | RPM | 3.9 | 470 | 120 |
| Biomass | BIO | 25.9 | 280 | 10.8 |
| Physically stabilized organic matter | POM | 94.8 | 11.3 × 10 ³ | 119 |
| Chemically stabilized organic matter | COM | 2565 | 12.2 × 10 ³ | 4.76 |
| Whole Soil Organic Matter | SOM | 91.7 | 24.3 × 10 ³ | 265 |

Jenkinson and Rayner, Soil Science 123, 6, 1977
S (kg) : Expected accumulation of organic matter after 10000 years when 1000kg ha⁻¹ of fresh organic matter is incorporated every year.
A₀ (kg) : Yearly gain of soil organic matter (kg ha⁻¹).
Calculated from S and mean age. A₀ = S/Average Age

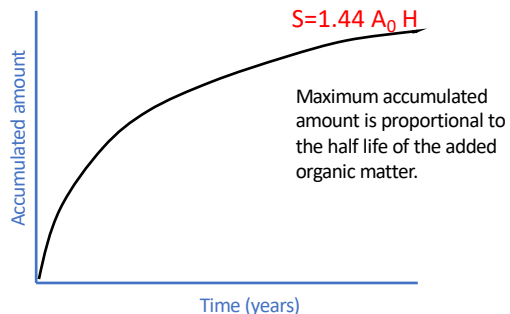
Accumulation of organic matter in soil

$$S = (1/\log_e 2) A_0 H$$

$$= 1.44 A_0 H$$

- S: Accumulated amount of organic matter after infinite years
- A₀: Annual input of organic matter
- H: Half life of organic matter
- 1.44H: Mean residence time

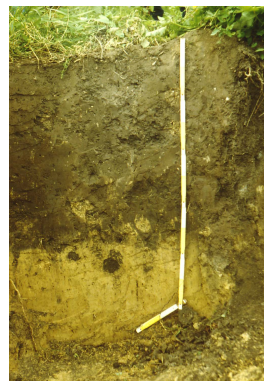
Accumulated amount of organic matter in soil approaches the maximum limit with time.



Carbon sequestration in soil

For the purpose of carbon sequestration, it is important to return the organic matter in stabilized form, for example, after composting, or after charring.

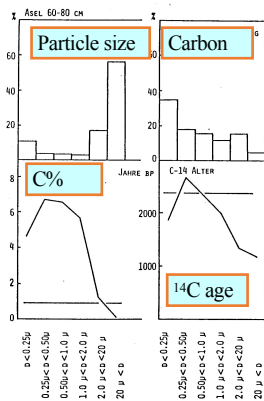
Black soil in Soellingen upland field



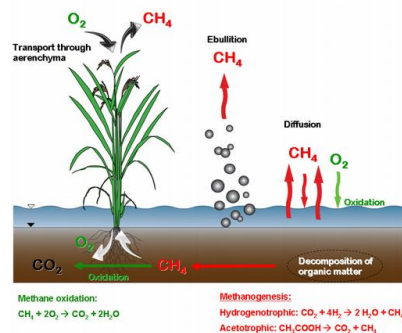
Soil organic matter stabilization on different size of soil particles

Organic matter bound to clay lasts long in soil

Scharpenseel, H.-W., Tsutsuki, K., Becker-Heidmann, P. and Freytag, J., Zeitschrift für Pflanzenernaehrung und Bodenkunde, 149: 582-597 (1986)



Formation of methane from paddy soils



Characteristics of paddy soils

- Characteristics of paddy soils are due to the flooding.
- Supply of oxygen is limited by the surface water, and the oxygen in the ploughed layer soil disappears. Iron oxide and manganese dioxide are consumed by the microbes and the soil becomes anaerobic.

Problems related to paddy soils (1)

- Problems due to soil reduction after flooding
 - Formation of volatile fatty acids
 - Acetic acid, Propionic acid, Butyric acid
 - Formation of hydrogen sulfide
 - due to sulfuric acid reducing bacteria
 - $\text{SO}_4^{2-} \rightarrow \text{H}_2\text{S}$

Problems related to paddy soils (2)

Formation of methane

- Around 10 % of the global methane formation is from paddy field.
- Formation of methane from paddy soils is controllable by field management, organic matter management, and irrigation water management.

Formation of nitrous oxide

- During the denitrification process, N_2O is formed.

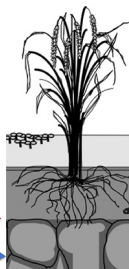
How to solve the problems

- Problems of volatile fatty acid, methane, and nitrous oxide formation can be solved by the following measures.
- Avoiding to bring the soil condition strictly anaerobic by conducting intermittent drying.
- Avoiding to incorporate fresh rice straw or fresh green manure.
- Wait some time after organic matter application before seeding rice.
- Refrain from excess nitrogen fertilizers.
- Apply ammonium form fertilizer deep in the reduced soil layer.

Composition of paddy soil

Paddy field has many excellent merits.

- Surface water
- Ploughed layer
- Ploughed pan layer
- Sub layer



Merits of paddy soil

- ① Problems due to continuous cropping are rare.
 - Reason
 - 1) Pathogenic fungi and nematodes die under anaerobic condition.
 - 2) Growth inhibiting substances are washed by the irrigation water.
- Rice cropping is continued for more than thousands of years in some places, e.g. rice terrace in Banaue, Philippines.

② Soil fertility does not decrease.

• Reason

- 1) Supply of nutrients from the irrigation water.
- 2) Decomposition of organic matter is repressed due to the anaerobic condition.
- 3) Various kinds of nitrogen fixing organisms are living in the surface water, and in the root zone soil.

③ Natural nutrients are supplied abundantly.

• Reason

- 1) Nitrogen is supplied from soil organic matter, and the formed ammonium is held by clay minerals and will not be washed away easily.
- 2) Iron phosphate becomes soluble after the reduced condition is formed.
- 3) Potassium and silicates are abundant in the irrigation water.
- 4) Soil pH becomes neutral after flooding the soil.

④ Due to the high ability to adjust the temperature, rice crop becomes tolerant to meteorological hazard.

• Reason

Due to the high specific heat of water, soil temperature is kept high and the cold injury of rice is mitigated in the cold area.

⑤ Removes nitrogen and phosphorus from the irrigation water.

• Reason

- 1) Excess nitrogen is denitrified.
- 2) Excess phosphorus is adsorbed on soil constituents.

⑥ Soil erosion hardly occurs.

• Reason

- 1) The paddy field is flat.
- 2) Soil erosion is controlled by the ridges and flooding water.

⑦ Weeds grow little.

- Major weeds in the paddy field are Eriochloa species and Carex species.
- Few weeds grow in flooded water.

⑧ Genetic potential of rice.

Rice plant tolerant to climate change.
New varieties from IRRI.

- Flood tolerant rice
- Drought tolerant rice
- Salt tolerant rice
- High temperature/ Low temperature tolerant rice
- Problem soil tolerant rice (Zinc deficiency • Potassium deficiency • Iron excess • Aluminum excess)