

# What soil profile survey tells you:

- What factor is limiting the plant growth (gravel, volcanic ash, clay, compaction of soil material, acidity, salt accumulation)
- Content and thickness of humus
- Drainage, water retention, dry or wet.
- Different soil layers composing the soil profile → History of soil

# Andosol profile in Obihiro Univ. Agr. & Vmed.



Kuroboku soil

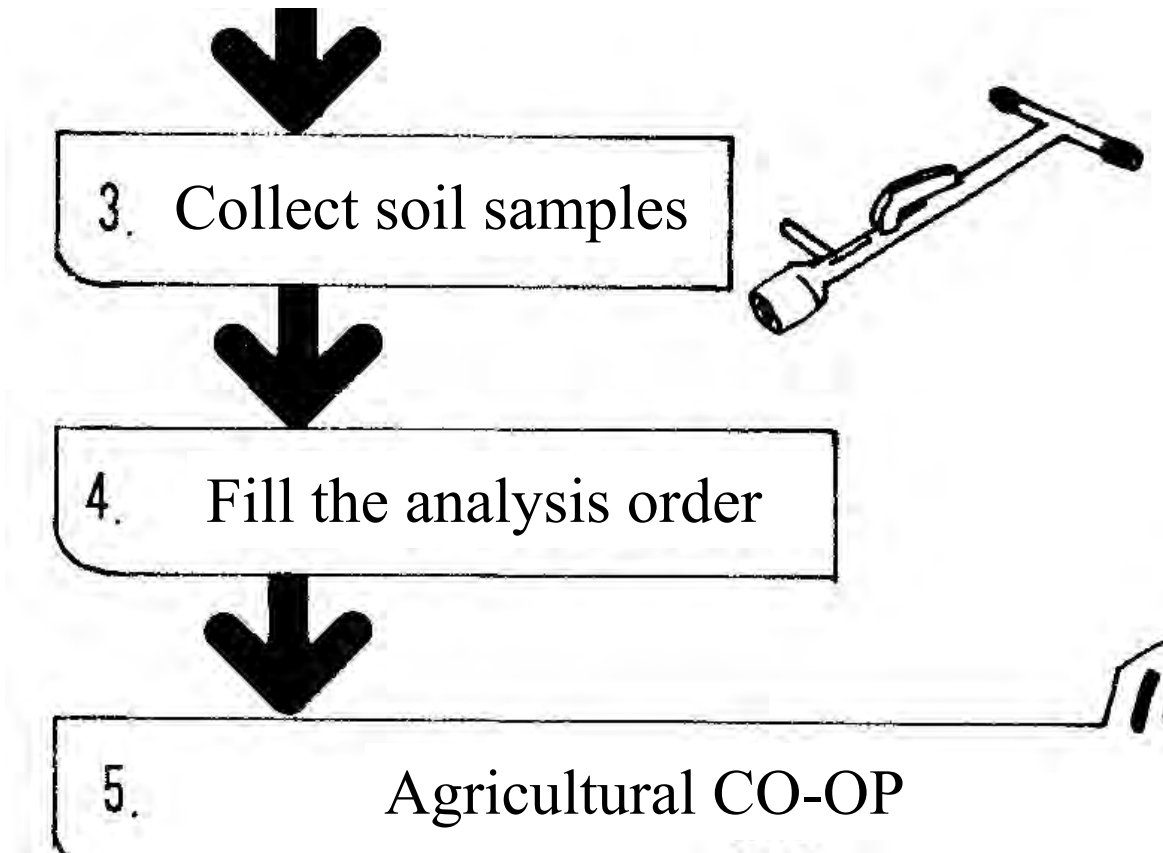
Plowed layer

Tarumae d  
volcanc ash

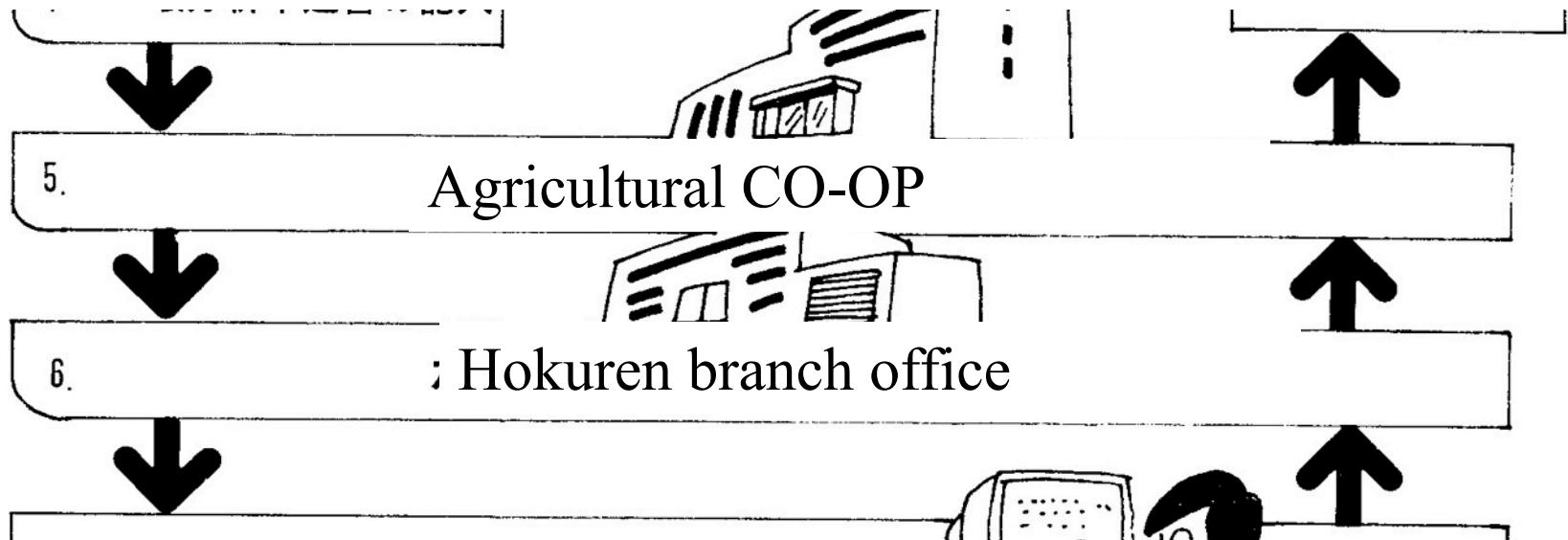
Eniwa loam

Alluvial soil

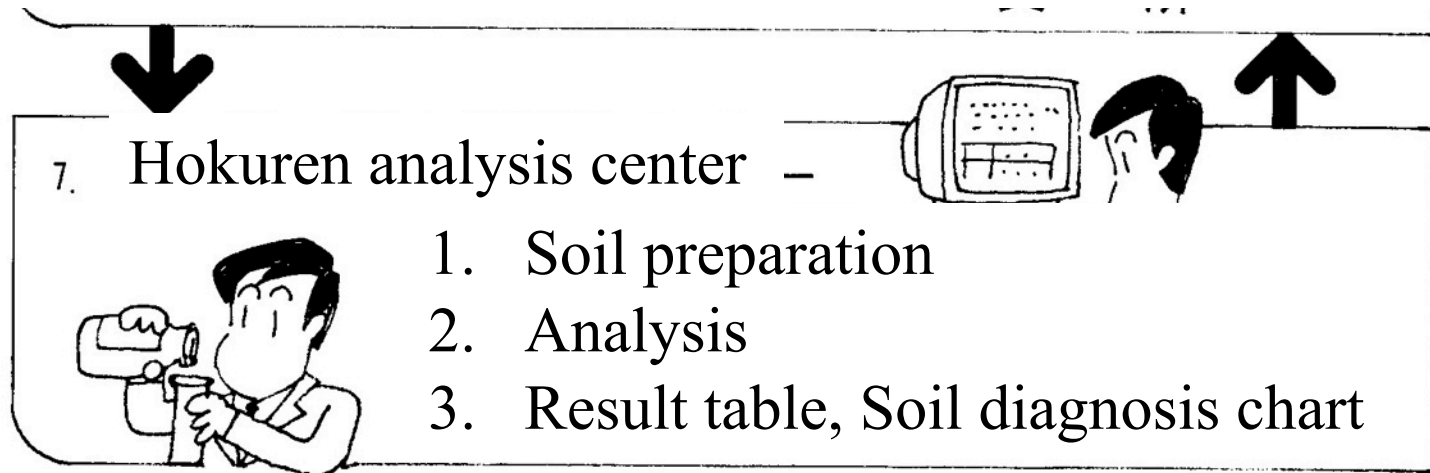
# Flow sheet of soil diagnosis 4



# Flow sheet of soil diagnosis 5

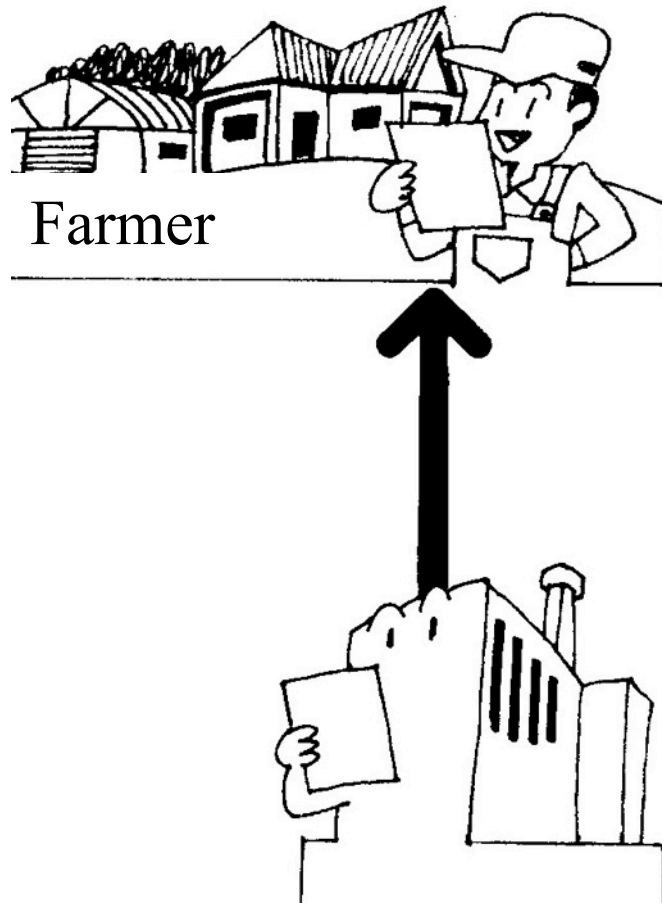


# Flow sheet of soil diagnosis 6



Tokachi Federation of Agricultural Cooperative  
Soil Analysis Laboratory

# Flow sheet of soil diagnosis 7



Guidance and advice to farmers according to soil diagnosis result

# Drying soil samples





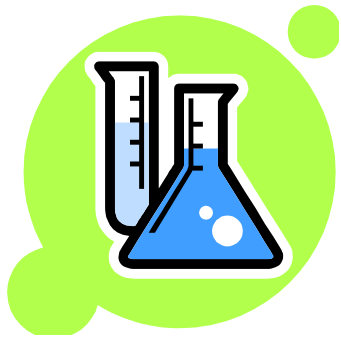
# Sieve soil samples (2mm)





# Soil samples after preparation





# Various Analysis Items and their significance



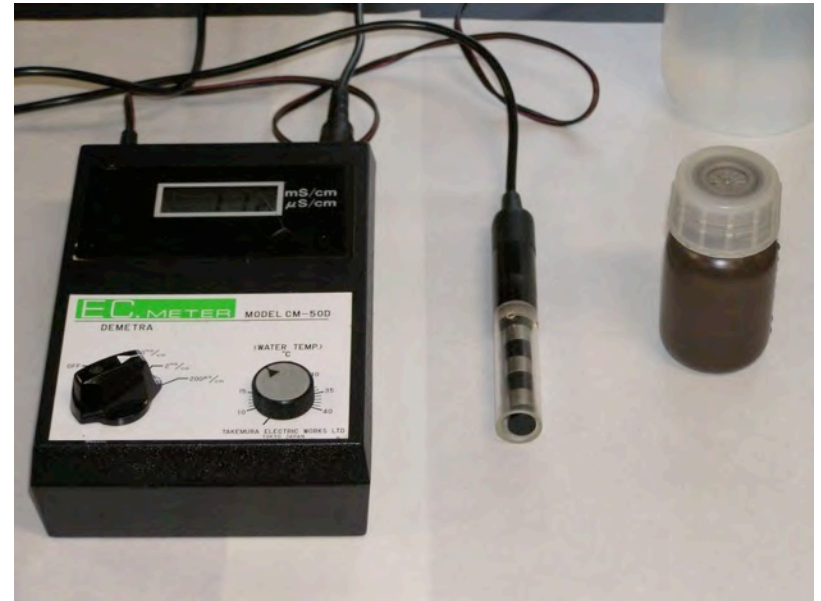
# pH(H<sub>2</sub>O)

- Concentration of free form H<sup>+</sup> in soil solution
- $\text{pH} = -\log(\text{H}^+)$
- Add 25 ml of water to 10g of soil.
- Shake 30 minutes.
- Measure the pH of turbid suspension using pH meter.

# Factors affecting soil pH(H<sub>2</sub>O)

- Fertilizer application
- Nutrient absorption by crops
- Seasonal change in climate, precipitation
- Partial pressure of CO<sub>2</sub>
- Activity of soil microbes
- Decomposition of soil organic matter
- Saturation degree of soil bases
- Leaching of soil bases
- Nitrification (NH<sub>4</sub><sup>+</sup>, NO<sub>3</sub><sup>-</sup>)

# pH meter & EC meter



# pH(KCl)

- Reflect the concentration of  $H^+$  and  $Al^{3+}$  adsorbed electrostatically to clay and humus.
- pH(KCl) decreases when degree of saturation by basic cations is low.
- Add 25 ml of 1 M KCl to 10g of soil.
- Shake 30 minutes.
- Measure the pH of turbid suspension using pH meter.

# Meaning of soil pH(KCl)

- Highly correlated with Al saturation degree of soil.
- pH(KCl) lower than 5.2 means
  - occurrence of exchangeable  $\text{Al}^{3+}$
  - Inhibition of plant growth by  $\text{Al}^{3+}$
- $\text{Al}^{3+} + \text{H}_2\text{O} \rightarrow \text{Al}(\text{OH})^{2+} + \text{H}^+$
- $\text{Al}(\text{OH})^{2+} + \text{H}_2\text{O} \rightarrow \text{Al}(\text{OH})_2^+ + \text{H}^+$



# pH(0.01M CaCl<sub>2</sub>)

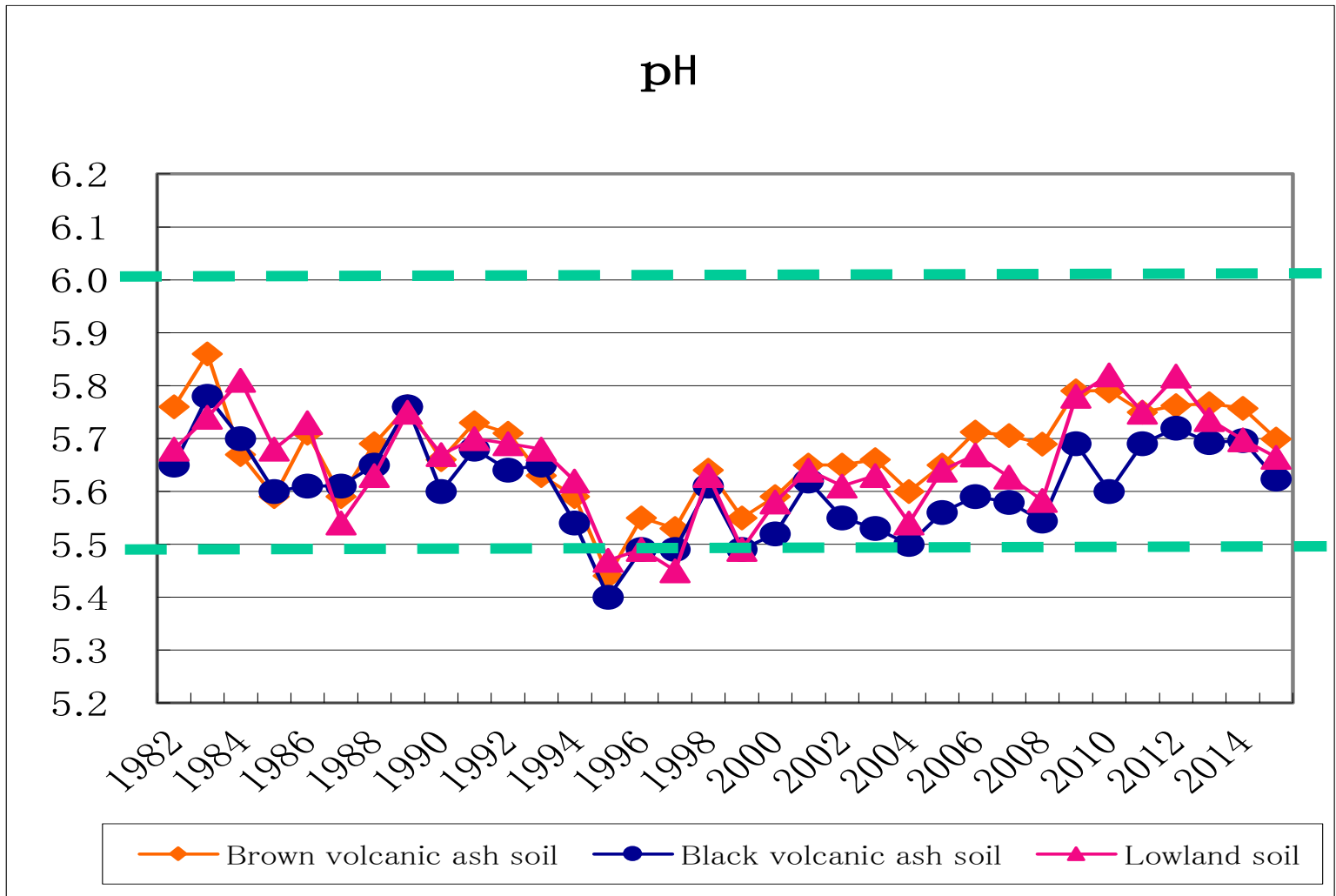
- Masking the effect of seasonal change and farm management
- To reflect the actual root zone environment more accurately, soil pH under dilute electrolyte concentration is more appropriate.

# Meaning of soil pH

< 5.0	Very acidic
5.0 – 5.5	Acidic
5.5 – 6.0	Weakly acidic
6.0 – 6.5	Slightly acidic
6.5 – 7.0	Neutral
7.0 – 7.5	Slightly alkaline
7.5 – 8.0	Weakly alkaline
8.0 – 8.5	Alkaline
8.5 <	Very alkaline

# Change in soil pH in Tokachi

Tokach Federation of Agricultural Co-operatives, Institute



# Effect of pH on plant growth

- $H^+$  ion inhibits the function of root ( $pH < 4$ )
- Increase in  $Al^{3+}$  ion (Inhibit growth at  $>1$  ppm level)
- Inhibit absorption of N, P, K, Ca, Mg, B, Mo and symptom of deficiency (in acidic range)
- Excess in Cu, Zn, Mn, Fe (in acidic range)
- Deficiency in Cu, Zn, Mn, Fe (in alkaline range)

# Exchangeable Acidity

- Weigh 10 g of air dried soil in to a flask or bottle.
- Add 25 mL of 1N KCl.
- Shake for 1 hour.
- Filter through a filter paper (Advantec No.6).
- Take 10 mL of the filtrate into a flask and titrate with 0.1 N NaOH.
- Consumed mL is multiplied by 12.5.
- Obtained value is  $Y_1$ .

# Electric conductivity (EC)

- Reflect total concentration of water soluble ions in soil solution
- Add 50 ml of deionized water to 10g of soil, shake 30 min. Measure EC of turbid suspension using EC meter.
- Unit is S/m, mS/cm or  $\mu\text{S}/\text{cm}$ , S: Siemens  
( $1\text{S}/\text{m}=10\text{ mS}/\text{cm} = 10^4 \mu\text{S}/\text{cm}$  )

# Meaning of soil EC

- High correlation with nitrate  $\text{NO}_3^-$  content
- Malnutrition under low EC ( $< 0.1 \text{ mS cm}^{-1}$ )
- Growth damage at high EC ( $> 1 \text{ mS cm}^{-1}$ )
- Adjust fertilizer application rate according to EC



# Greenhouse soil diagnosis according to pH and EC

pH(H <sub>2</sub> O)	7.0	Excess Ca → Apply sulfate fertilizer	Excess fertilizer → No fertilizer, Remove salts by flooding
	5.5	Insufficient fertilizer → Apply fertilizer and organic matter	Excess N fertilizer → Frequent Watering, Remove salts by flooding
		Suitable	
		0.4	1.0
		EC (mS/cm)	

# Application rate of basal fertilizer (N, K) according to soil EC (dS m<sup>-1</sup>) in upland field

Soil Type	< 0.3	0.4-0.7	0.8-1.2	1.3-1.5	1.6 <
Humic andosoil	Standard rate	2/3	1/2	1/3	No fertilizer
Sandy Fine textured	Standard rate	2/3	1/3	No fertilizer	No fertilizer
Sand dune/ immature	Standard rate	1/2	1/4	No fertilizer	No fertilizer

# Humus

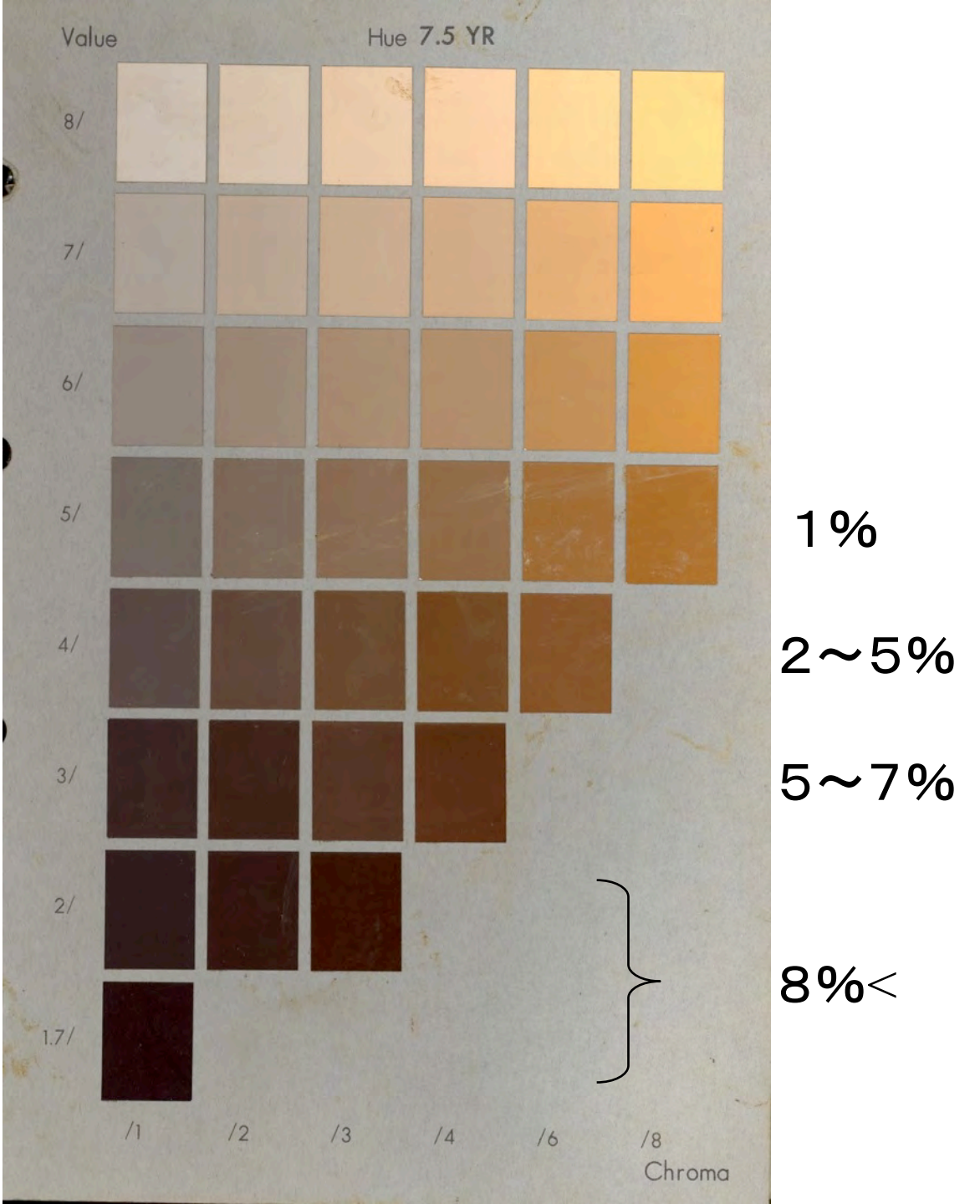
- Humus = Soil organic matter

## Method of determination

- Rapid estimation by soil color
- Tyurin method (Potassium dichromate oxidation/ Titration)
- Dry combustion method (Instrumental analysis)

# Standard Soil Color Chart Hue 7.5YR

Relationship  
between soil  
color and organic  
matter content

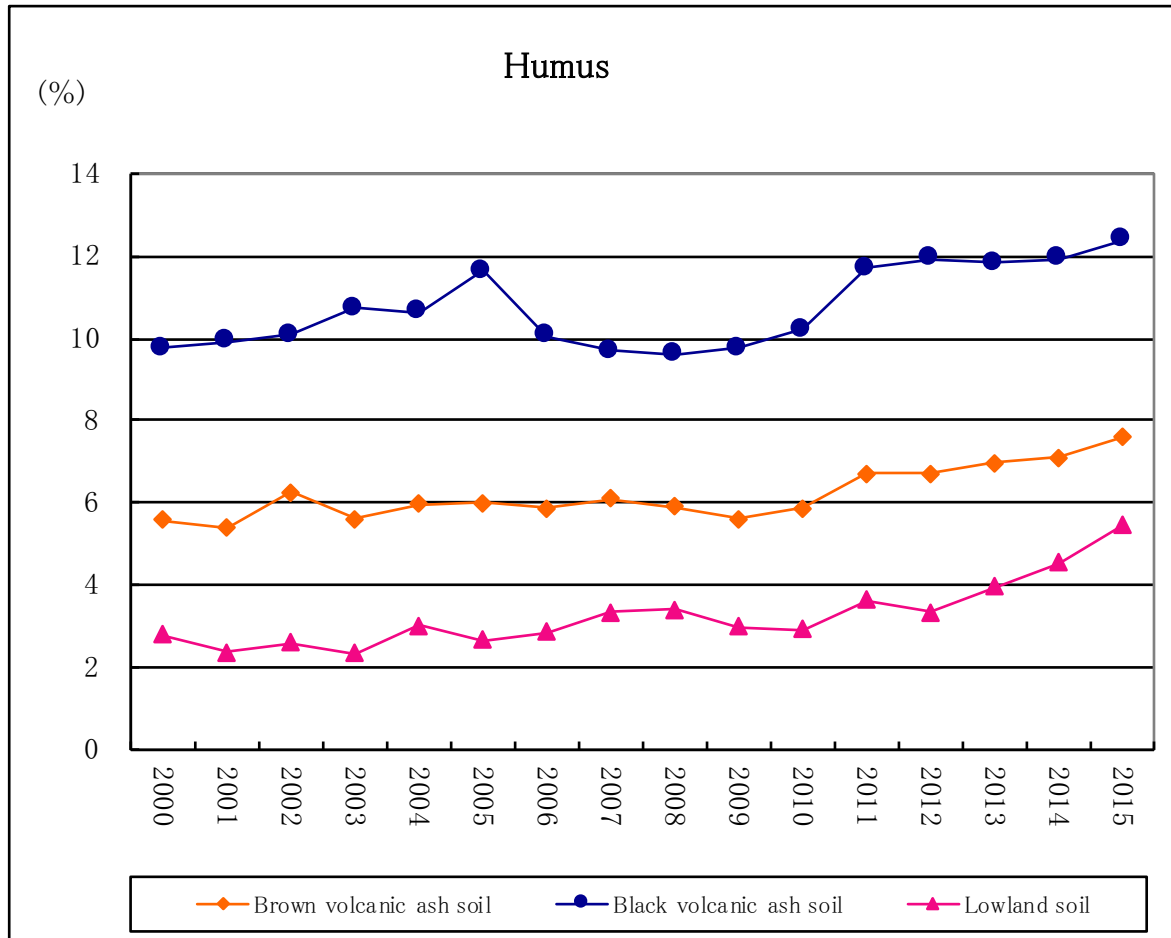


# Importance of humus

- Soils with high humus content are generally fertile and easily manageable.
- Exception → Andosol (Kuroboku in Japan)
- Supply nutrients (especially N)
- Hold soil moisture
- Hold nutrients (Cation Exchange Capacity)
- Formation of Soil Aggregate Structure

# Change in soil humus in Tokachi

Tokach Federation of Agricultural Co-operatives, Institute



# Nitrogen Analysis

- Nitrogen is the most important constituent of fertilizer.



# Inorganic nitrogen

- Ammonium nitrogen

Extracted by 1N KCl, 2N KCl

- Nitrate nitrogen

Extracted by Water, 1N KCl, 2N KCl

- Determine by steam distillation/ titration or colorimetry
- Rapidly available to crops

# Available nitrogen

- Potential amount of inorganic nitrogen formation
- After incubating 4 weeks at 30 °C, total amount of formed inorganic nitrogen is determined.
- Incubation under upland or paddy condition.
- Problem: Time consuming method

## Phosphate buffer (pH7) extraction method (Rapid estimation method for available nitrogen)

- Extracted nitrogen content or absorbance at 420 nm of the extracted solution showed high correlation with available nitrogen estimated by incubation method.

# Hot water extractable nitrogen

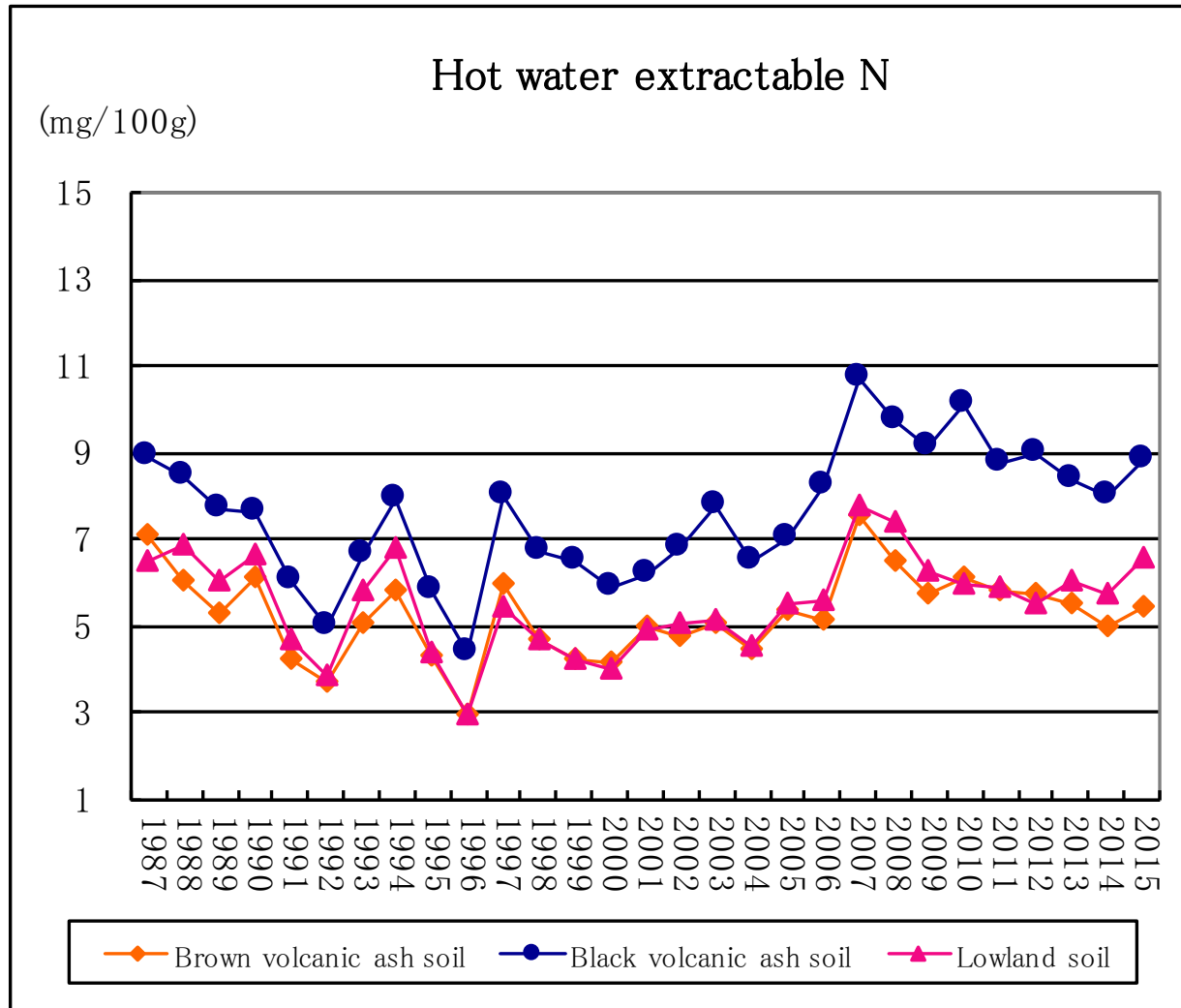
- Another measure of available nitrogen
- Soil + water (1:10)
- Autoclaved (105 °C/modified to 121 °C, 1 hour)
- Filtered
- Extracted solution is digested by Kjeldahl method
- Nitrogen is determined by colorimetry

## Adjustment of N application rate according to hot water extractable nitrogen

<b>Hot water N (mg / 100 g)</b>	<b>N application rate (kg / 10 a)</b>
1, 2	24
3, 4	20
5, 6	16
7, 8	12
9, 10	8
Higher than 11	8

# Change in hot water extractable N in Tokachi

Tokach Federation of Agricultural Co-operatives, Institute



# Total Nitrogen

- Kjeldahl digestion (conc  $\text{H}_2\text{SO}_4 + \text{K}_2\text{SO}_4 +$   
Catalyst(Cu, Hg, Se) )



- Instrumental (Dry combustion method)
- C/N is calculated
- C/N is related to the pattern and rate of nitrogen mineralization

# Kjeldahl digestion apparatus





# Release of N from organic matter applied to soil every year

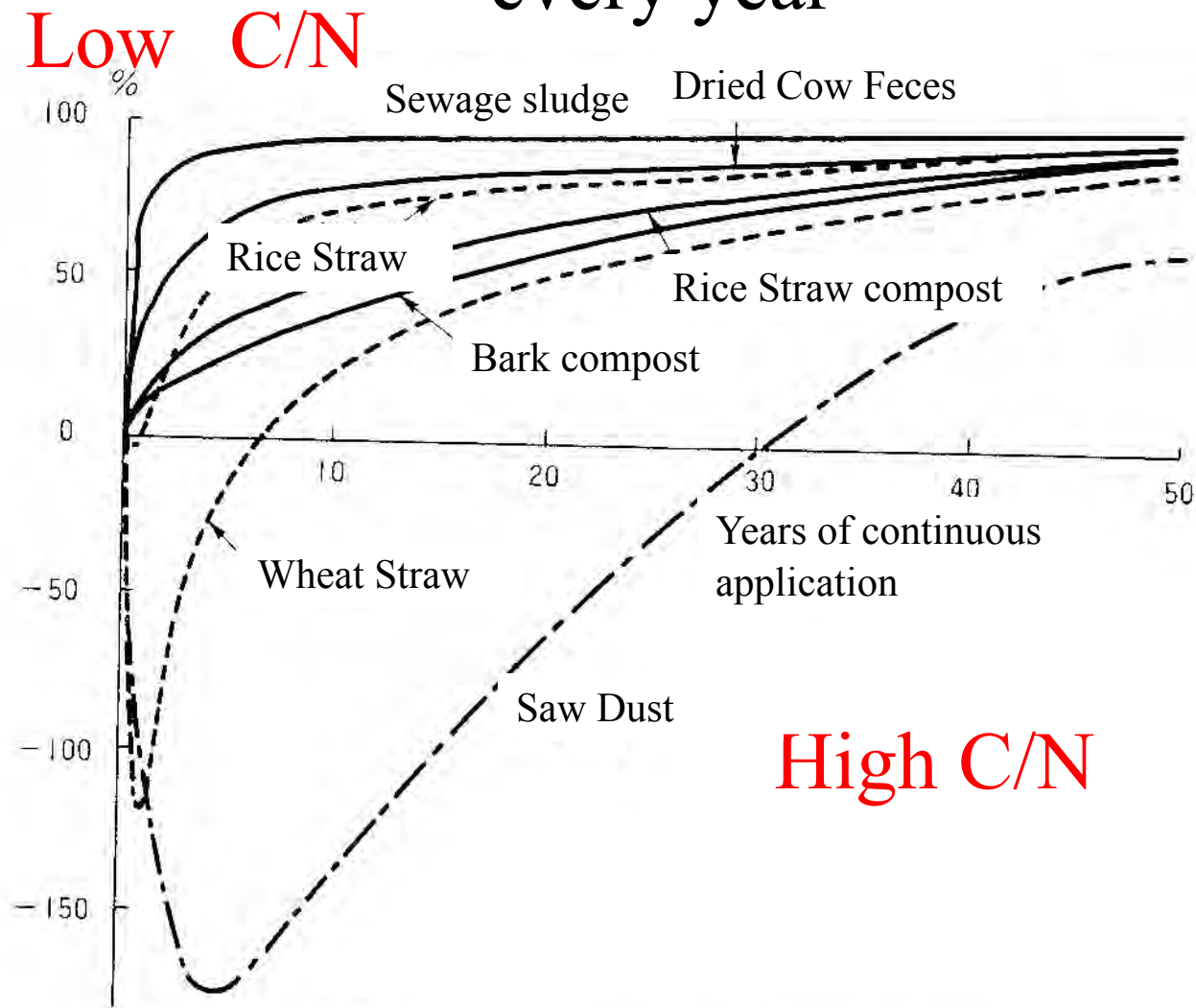


図10 有機物連用の場合の炭素の集積率, 毎年の窒素の放出率の予測値  
1年間に添加する量を100とした場合 (志賀ら, 1985)